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(54) A METHOD OF PREPARING AN ANTI-FRICTION MATERIAL

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The present invention relates to a method 25 of preparing an antifriction material.

Said materials are useful for the production of friction couple elements such as sealing rings, thrust bearings, supporting jour-nals, sliding bearings for electric motors 30 as employed in oil wells; chemical equipment such as centrifuges and pumps for corrosive liquids, e.g. acids, kerosene, petroleum, oils, salt solutions, alkali solutions, and other liquids which are corresive 35 at elevated temperatures.

Known in the art is an antifriction material containing silicon carbide and carbon; this antifriction material is prepared using carbonaceous materials as the starting components.

As a starting component for the production of said known material use is made of a graphite blank which is treated with silicon oxide vapours at a temperature of from 1,600 to 2,200°C. The silicon oxide vapours are evolved from silicon oxide atomized in a hydrogen current. The use of a reducing atmosphere and elevated temperatures contributes to the formation of silicon vapours and interaction thereof with graphite. As a consequence, silicon carbide is formed which, being deposited on the

graphite surface, produces a superficial laver.

(11)

A material is thus obtained which contains silicon carbide and graphite. Such material, however, has some essential disadvantages. First of all, it possesses different thermal expansion coefficients of graphite and silicon carbide, which results in the formation of cracks either in the graphite or in the silicon carbide during temperature variations. Therefore, when this known material is employed under oxidizing conditions in the presence of an oxidizing medium at temperatures of from 300 to 600°C, it is liable to break down, since such medium penetrates into cracks and destroys graphite disposed under the silicon carbide layer, whereby the antifriction properties of the whole material are impaired.

Moreover, the complete lack of graphite possessing lubricating properties in the superficial layer of the material results in an increased coefficient of friction, whereby the antifriction properties of the material to be obtained are also deteriorated.

It is an object of the present invention to obviate or mitigate the disadvantages of the aforesaid antifriction materials.

According to the present invention there is provided a method of preparing an antifriction material, comprising forming a mixture of carbon and a binder, compressing the mixture at a temperature within the range from 150 to 180°C to produce a blank having a density of 1 to 1.4 g/cm^a, heating the blank at a temperature within the range from 800 to 1000°C, and impregnating the heated blank with silicon at a temperature within the range from 1700 to 2050°C.

This antifriction material produced by the present method is stable under frequent heat variations and retains its properties under oxidation conditions at elevated temperatures.

Preferably the binder is thermosetting

Preferably the carbon is selected from raphite powder, carbon black, carbon fibres or mixtures thereof.

In order to completely remove gas from the pores of the material being obtained.

[Price 33p]

it is preferred that impregnation with silicon be performed in vacuum.

In order to increase the plasticity of the material being obtained, the impregnation with silicon may be effected in the presence of nickel, cobalt, zirconium, niobium, titanium, molybdenum, tungsten, tantalum or chromium, taken separately or in admixture.

It is advantageous, that during the mixing of the above carbon materials with the binder, iron and aluminium be mixed thereto either separately or in admixture.

The present invention will be better understood from the following detailed description of a specific example of an embodiment.

An antifriction material comprising silicon carbide in an amount of from 20 to 65% by weight, carbon in an amount of from 20 75 to 10% by weight, and silicon in an amount of from 5 to 25% by weight is prepared from the starting components as follows; as such components use may be made of carbon containing materials selected from graphite powder, carbon black and carbon fibers. Said materials may be em-

ployed both separately and in admixture with each other.

When graphite powder is used the antification properties of the material being pre-

pared is increased; when carbon black is used the silicon carbide content in the material being prepared is increased and this enhances its wear-resistance and when carbon fibers are introduced the resilience

of the material is increased.

The starting carbon containing materials are mixed with a binder selected from the group of thermosetting resins such as phenolformaldehyde resins possessing a property of softening upon heating over 100°C and

of softening upon heating over 100°C and of wetting carbon particles, thus imparting necessary plasticity of the whole composition.

The mixture of carbon and binder is compressed at a temperature within the range from 150 to 180°C to form a plasticized blank which is further compressed until its density becomes 1.0 to 1.4 g/cm³.

As a consequence, a porous carbon containing material is obtained and its pores are then filled with silicon. A density of less than 1.0 g/cm³ results in the production of a carbon material possessing rather low strength for the subsequent impregnation with silicon. The use of a density of above 1.4 g/cm³ causes a decrease in the amount of pores and, hence, an incomplete impregnation of the material with silicon.

The resulting blank with the abovementioned density is then heat-treated according to the following scheme: first calcined in an inert atmosphere at a temperature from 800 to 1,000°C and then impreg-

nated with molten silicon at a temperature of from 1,700 to 2,050°C.

The carbon containing blank may be impregnated in vacuum. In doing so, gases are removed from the pores of the material being treated, and the impregnation proceeds more intensively.

However, the siliconizing of a carbon containing blank may be also performed in an inert atmosphere, but in this case a material is produced which has a somewhat lower silicon carbide content.

During the intermixing of the carbon containing material and the binder, powders of iron and/or aluminium may be added to the mixture in such a manner that their respective content in the final material is up to 3% by weight.

Iron facilitates the impregnation of porous graphite with silicon, while aluminium contributes to the densification of the carbide phase, thus resulting in the production of a gas-impermeable material.

Said additives may be incorporated into the composition either simultaneously or separately.

For a better understanding of the present invention some specific examples of its embodiment are given hereinbelow.

EXAMPLE 1

A powder of artificial graphite having fineness 0—1.25 mm is mixed with a phenolformaldehyde resin the components being taken in the proportions of 85 and 15% by weight respectively. The resulting mixture 100 is compressed, upon heating at 150°C and under the pressure of 120 kg/cm² to a blank with the density of 1.25 g/cm² which is then calcined in an inert atmosphere at a temperature of 800°C. The calcined blank 105 is subjected to a high-temperature treatment in molten silicon at a temperature of 2,050°C. The resulting material has the following composition:

silicon carbide 25% by weight; graphite 50% by weight; free silicon 25% by weight.

The material thus prepared has the follow- 115 ing properties:

density 2.25 g/cm³;
ultimate compression
strength 1,300 kg/cm²; 120
ultimate bending
strength 700 kg/cm²;
coefficient of friction
resilience 2 kgf.cm/cm².

Example 2

A powder of graphite having fineness of 0—0.2 mm. is mixed with a phenol-formaldehyde resin, the components being em-

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	ployed in the proportions of 85 to 15% by weight respectively. The resulting mixture is compressed at a temperature of 160°C	at a temperature of 1,700 C. The resulting material has the following composition:	-
5	which is then calcined in an inert atmosphere at the temperature of 900°C, treated in molten silicon at the temperature of 1.750°C	silicon carbide	70
10	The resulting material has the following composition:	and has the following propperties: density 2.70 g/cm ² ;	
	silicon carbide 65% by weight; graphite 28% by weight; free silicon 7% by weight.	strength 4,200 kg/cm ² ; ultimate bending	75
15	The material has the following properties:	strength 1,000 kg/cm²; coefficient of friction 0.05; resilience 3.2 kg.cm/cm².	80
2 Ò	density ultimate compression strength ultimate bending strength coefficient of friction resilience 2.55 g/cm³; 3,000kg/cm²; 900 kg/cm²; 0.05; 2.8 kgf.cm/cm².	EXAMPLE 5 Graphite powder having fineness of 0—0.5 mm. and aluminium powder having fineness	85
25	Example 3 A composition prepared by mixing graphite powder with a fineness of 0 to 0.5	a temperature of 150°C and under a pressure of 200 kg/cm³ into a blank having a density of 1 to 1.4 g/cm³ and the blank is calcined in an inert atmosphere at a temperature of 150°C and under a pressure of 150°C and u	90
30	mm. with a phenol-formaldehyde resin in proportions of 85 and 15% by weight respectively, is compressed at the temperature of 170°C and under the pressure of 200 kg/cm ² to a blank having a density of	in modern silicon at a temperature of	95
35	1.3 g/cm ³ which is calcined in an inert atmosphere at the temperature of 900°C. The resulting blank is treated in molten silicon at the temperature of 1,800°C. The material thus prepared has the following composition:	silicon carbide 72% by weight; graphite 25% by weight; free silicon 3% by weight, 1 and has the following properties:	1 00
40	silicon carbide 47% by weight; graphite 48% by weight; free silicon 5% by weight.	density 2.70 g/cm²; ultimate compression 1 strength 4,500 kg/cm²; ultimate bending	105
45	and has the following properties: density 2.40 g/cm²:	strength 1,200 kg/cm ² ; coefficient of friction 0.04;	110
50	ultimate compression strength 3,200 kg/cm²; ultimate bending strength coefficient of friction resilience 900 kg/cm²; 0.05; 2.8 kgf.cm/cm².		115
55	Example 4 Graphite powder having fineness of 0—0.5 mm. and iron powder having fineness of 0—0.05 mm. are thoroughly mixed with a	perature of 150°C and under a pressure of 300 kg/cm ³ into a blank having a density 1 of 1 to 1.4 g/cm ³ and the blank is calcined	20
60	phenol-formaldehyde resin in proportions of 82, 3 and 15% by weight respectively. The resulting mixture is compressed at a temperature of 150°C and under a pressure of 150 kg/cm ² into a blank with a density of	in an inert atmosphere at a temperature of 900°C. Cylindrical samples made of this blank are siliconized at a temperature of 2,050°C. The resulting material has the 1 following composition:	25
65	1.4 g/cm³ which is further calcined in an inert atmosphere at a temperature of 900°C. Then the blank is treated in molten silicon	silicon carbide 71.3% by weight; graphite 22.8% by weight; free silicon 5.9% by weight 1	30

and features the following properties: impregnated with an alloy containing 75% by weight of silicon and 25% by weight of density 2.70 g/cm3; nickel at a temperature within the range from 1700 to 2050°C. The samples are thereafter tested for resilience. The resilience ultimate compression strength 4,300 kg/cm2; ultimate bending is 12 kgf.cm/cm². strength 1,100kg/cm²; coefficient of friction 0.04. WHAT WE CLAIM IS:-1. A method of preparing an antifric- 60 tion material, comprising forming a mixture 10 EXAMPLE 7 of carbon and a binder, compressing the mixture at a temperature within the range from 150 to 180°C to produce a blank having a density of 1 to 1.4 g/cm³, heating the blank at a temperature within the range from 800 to 1000°C, and impregnating the heated blank with gilleng at a temperature A powder of artificial graphite with a fineness of 0-0.315 mm., carbon fibres, and a phenol-formaldehyde tesin taken in weight proportions: graphite, 80%; carbon fibres, 5%; and resin 15%, are thoroughly intermixed. The resulting mixture is compressed at a temperature of 150°C and under a presheated blank with silicon at a temperature within the range from 1700 to 2050°C. sure of 100 kg/cm² into samples having a 2. A method as claimed in claim 1, 70 density of 1 to 1.4 g/cm² and dimensions (cross-section): 10×10 mm., and 120 mm. wherein impregnation is effected in a vacuum. long. Compressed samples are calcined at a temperature of 900°C and then siliconized 3. A method as claimed in any one of the preceding claims, wherein the carbon is at a temperature of 2,050°C. The samples are thereafter tested for resilience. The preselected from graphite powder, carbon black, 75 carbon fibres or mixtures thereof. sence of fibers in the samples has improved 4. A method as claimed in any one of their resilience by three times as compared the preceding claims, wherein the impregnato similar samples made of the known tion with silicon is effected in the presence material. During the tests the following proof nickel, cobalt, zirconium, niobium, perties have been shown: titanium, molybdenum, tungsten, tantalum, or chromium, or mixtures thereof. density ultimate compression 2.45 g/cm3; 5. A method as claimed in any one of the preceding claims, wherein aluminium strength 4,000 kg/cm²; and/or iron is added to the mixture during ultimate bending mixing. strength 1,000 kg/cm²; 6. A method of producing an antifriction coefficient of friction 0.05 material according to claim 1, substantially resilience 10 kgf.cm/cm². as hereinbefore described and with reference to any one of the Examples. 40 Example 8 7. An antifriction material whenever prepared by the method claimed in any one Graphite powder with a fineness of 0-0.315 mm., carbon fibres and a phenolof claims 1 to 6. formaldehyde resin taken in weight proportions; graphite, 70%; carbon fibre, 15%; resin, 15% are thoroughly intermixed. The FITZPATRICKS. Chartered Patent Agents, resulting mixture is compressed at a tem--18 Cadogan Street perature of 180°C and under a pressure of 50 kg/cm² into samples having a density Glasgow, G2 6QW, and of 1 to 1.4 g/cm² and dimensions: 10 × 10 × 120 mm. The compressed samples are calcined at a temperature of 900°C and then Warwick House, Warwick Court,

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